Effect of Heat Treatment Parameters on Microstructure and Mechanical Properties of (AISI1005) Steel

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Abstract
This study aims to investigate tensile properties and work hardening behavior of dual phase (DP) steels. A series of DP steels containing ferrite and martensite with different volume fractions of martensite (Vm) were produced by intercritical heat treatment. Microstructural investigations, hardness test, and tensile test were carried out. The experimental results showed that dual phase steels at 760°C and 10s have excellent mechanical properties in terms of tensile strength, ductility and fracture energy. A further increase in Vm was found to increase tensile strengths and ductility. The increasing and then decreasing trend in tensile strength is in contrast to the law of mixture. These unusual behaviors are discussed and explained. Work hardening behavior was analyzed in terms of Holloman analysis.

Keywords Dual phase steel, heat treatment, mechanical properties.

1. Introduction

1-1 Advanced High-Strength Steel (AHSS)
AHSS: A series of high-strength steels containing microstructural phases other than ferrite and pearlite, these other phases include martensite, bainite, and/or retained austenite in quantities sufficient to produce unique mechanical properties. Most AHSS have a multi-phase microstructure [1]. AHSS can be classified according to the microstructure including dual phase (DP), transformation induced plasticity (TRIP) assisted, complex phase (CP) and martensitic (MS) steels.

Fig (1) compares the tensile strength and total elongation of lower strength steels [interstitial free and mild steel], conventional high strength steels (carbon–manganese, bake hardenable and high strength low alloy steels) and AHSS [Y,3].

Fig. 1 - Relationship between ultimate tensile strength and total elongation (50.8 mm gauge length) for various types of steel [1].
1-2 Dual Phase Steel

Ferrite–martensite dual phase (DP) steel is one of the most common Advance High Strength Steels (AHSS) which is currently used in automotive industry [4]. Usually ferrite–martensite DP steels are produced by intercritical annealing followed by rapid cooling. During the intercritical annealing small pools of austenite are formed in the ferrite matrix, which subsequently transform into martensite upon rapid cooling. Intercritical heat treatment is the simplest way to enhance low alloys (carbon content less than 0.2%) steels to dual phase microstructure with superior strength–ductility combination [3].

In this combination of two phases, martensite contributes with high strength and ferrite matrix provides good elongation that can produce a good combination of strength and ductility for applications which required good formability. This unique composite microstructure offers other interesting mechanical properties such as continuous yielding, low yield stress to tensile strength ratios and high initial work-hardening rate [2, 5].

1-3 Application of Dual Phase Steel

Fig. 2 shows the use of modern multiphase steels for structural components, taking the example of a new auto body [6]. Honda and BMW as shown in Fig. 3, two car manufacturers that most quickly and warmly welcomed DP grades into their portfolios, have each become known for their forward-thinking use of materials [7]. The wheels in the ULSAB-AVC are made of 1.2 mm DP 350/600 steel and the inner portion is made of 1.8 mm HSLA 490/600, resulting in an optimized, lightweight steel wheel design. The disc is made of 2.1 mm DP 500/800 as shown in Fig. 4. [8].

Fig. 2: Application of multiphase steels to the body structures of VW Touareg and Porsche Cayenne [6].
2. Experimental procedure

The steel used in the present investigation was 2-mm in thickness AISI 1005 sheet steel whose chemical composition is given in Table 1. Fig. 5 shows the engineering stress–strain curve and Fig. 6 shows the microstructure of base metal. The microstructure of the base metal sample consists of mainly ferrite with small amount of pearlite.

<table>
<thead>
<tr>
<th>Element %</th>
<th>C</th>
<th>Mn</th>
<th>P</th>
<th>S</th>
<th>Si</th>
<th>Cr</th>
</tr>
</thead>
<tbody>
<tr>
<td>AISI1005</td>
<td>0.048</td>
<td>0.58</td>
<td>0.03</td>
<td>0.02</td>
<td>0.028</td>
<td>0.02</td>
</tr>
</tbody>
</table>
For this study, the $AC_1$ and $AC_3$ temperatures were calculated to be 735 and 856 °C, respectively by Eqs. (1) and (2) [9].

$$AC_1 (°C) = 751 - 16.3C - 27.5Mn - 5.5Cu - 5.9Ni + 34.9Si + 12.7Cr + 3.4Mo$$  \hspace{1cm} (1)

$$AC_3 (°C) = 881 - 206C - 15Mn - 26.5Cu - 20.1Ni - 0.7Cr + 53.1Si + 41.7V$$  \hspace{1cm} (2)

The heat treatment which is used in this work is shown in Fig. 7. All of the specimens were heated in intercritical temperature range (between $AC_1$ and $AC_3$). All the specimens were held between (5-20) min in (CARBOLITE) furnace as shown in Fig (8) and followed by water quenching (WQ). The heat-treated samples are designated by capital letter, as shown in Fig (7). Metallographic specimens were prepared according to the standard procedure from the base metal and heat-treated samples and etched with 2% nital solution. Microscopic examinations were carried out by an optical microscope.

The tensile test was conducted on base metal and heat-treated specimens with 50mm gage length, 200mm overall length, 20mm width of grip section and 12.5mm width of reduced section using an (QUASAR 25) testing machine as shown in Fig (9) with strain rate of 10mm/min at room temperature in accordance to ASTM E8M [10].

Key parameters obtained from stress–strain curves include yield strength ($\sigma_Y$), ultimate tensile strength (UTS), uniform elongation (UEL), total elongation (TEL) and tensile fracture energy. The amount of energy absorption was digitally calculated by measuring the area under the stress–strain curve up to the final fracture point using the concept of Riemann Sums Eq. (3) [11].

$$\int_0^{L_{\text{max}}} F \, dx = \sum_{n=1}^{N} F(n) \cdot [L(n) - L(n - 1)]$$  \hspace{1cm} (3)

Where $F$ is the load, $X$ the displacement, $L_{\text{max}}$ the displacement at the peak load, $n$ the sampled data and $N$ the peak load.
3. Results and discussion
3.1. Microstructural investigation
The microstructures of the heat-treated samples (A, B and C) are shown in Fig. 10. Increasing \( \text{Vm} \) is clearly observed in Fig. 10a–c. This complies with lever rule in the ferrite–austenite dual phase region. According to the lever rule, increasing the temperature increases the austenite volume fraction, which then will transform to martensite upon quenching in the water. It worth noticing that by increasing the temperature, \( \text{Vm} \) increases at higher rate.

Fig. 11 shows variations hardness of the heat-treated samples with increasing heat treatment temperature. Hardness of base metal steel is (134) HV. Hardness of the heat-treated samples
are higher than the hardness of the specimen in the as-received condition. The higher hardness of the dual phase steels due to the presence of the martensite phase.

![Image of microstructures](image_url)

**Fig. 10.** The microstructures of the heat-treated samples at various intercritical temperatures of (a) 760 °C (b) 800 °C and (c) 820 °C. [80X]

![Image of hardness graph](image_url)

**Fig. 11.** Hardness for heat-treated samples (A, B, C, D) and Base metal.

### 3.2. Tensile properties

3.2.1. Continues yielding behavior

The stress–strain curve of the base metal sample has yield point elongation and exhibit well-defined yield point (see Fig. 6). In low carbon steels, this behavior has been attributed to the effect of interstitial solute atoms (carbon), called the Cottrell atmosphere, on locking-in the dislocation. When stress is applied to such steel in a tensile test, it must exceed a certain critical value to unlock the dislocations. The stress necessary to move the dislocations is less than the stress required to unlock them; hence the phenomenon of a sharp yield drop and the appearance of an upper and lower yield point in the tensile stress–strain curve [12].

The yield point has come to be accepted as a general phenomenon, since it has been observed in a number of other metals and alloys. Three factors which control sharp yield point in materials are as follows [13,14]:

1) Low density of mobile dislocation in initial stage of deformation,
2) Rapid dislocation multiplication during plastic deformation and,
3) Low stress dependence of dislocation velocity.
However, this phenomenon is not observed for dual phase steels. As can be seen from Fig. 12 the stress–strain curves of dual phase steels exhibit continuous yielding behavior. The continues yielding of the ferrite–martensite dual phase has been related to the following [15–16]:

1) Presence of unpinned dislocations which are created in a ferrite matrix via plastic deformation during the austenite to martensite transformation.
2) These unpinned dislocations, located at the ferrite–martensite boundaries are assumed to be mobile in the early stage of plastic deformation.

![Continuous yielding behavior](image)

**Fig. 12. Engineering stress–strain curves for heat-treated samples (A, D, E).**

### 3.3. Work hardening

The flow behavior of the most metals and alloys can be described by Hollomon Eq. (4) [17]:

$$\sigma = Ke^n$$  

Where ‘K’ and ‘n’ are constants which are normally called: the strength coefficient and strain hardening exponent, respectively. Drawing the stress–strain data in logarithmic scale and fitting a line to these data, determines the values of these coefficients for the under investigation metal. Stain hardening exponent (n) is the slope of this line and its cross section with ε = 1 (log ε = 0), gives the strength coefficient (K).

Stain hardening exponent (n) is a good indicator for work hardenability of the material.
3.4 Fracture energy

Fig. 14 shows the effect of temperature of heat treatment on the fracture energy of DP steels. In the automotive industry, the energy absorption capability is an important parameter in vehicle crashworthiness. Variation of fracture energy follows a similar trend for variation of elongation with \( V_m \). The failure energy increases and maximum failure energy in stage (B). Fracture is understood in terms of the nucleation (either by particle cracking or interfacial decohesion), growth, and coalescence of damage [19]. As the martensite volume fraction increases, its flow stress decreases, resulting in an increase in the potential for martensite to co-deform with the ferrite matrix. In these cases, the nucleation of voids is much more difficult, with the final result being that the fracture strain and fracture energy are increased [19]. It is also interesting to note that 4\% volume expansion accompanies the transformation of austenite to martensite. This volume expansion creates a compression stress in the ferrite/martensite interface which impedes the crack growth even if it is nucleated [20]. This can contribute to the increasing energy absorption capability. With further increase in \( V_m \), the matrix structure changes from ferrite to martensite and coherency of the ferrite matrix is also damaged by high amount of martensite phase, thus the DP structure becomes brittle and the failure energy is reduced.

Fig. 14: The variation of the fracture energy with temperature of heat treatment
4. Conclusions
1 – A different samples of dual phase steels containing ferrite and martensite with different volume fractions of martensite (Vm) was produced by intercritical heat treatment.
2 - It was found that martensite volume fraction increases by increasing the intercritical heat treatment temperature.
3 - Generally, dual phase steels in stage A (780 C, Time 10 S) exhibit the optimum mechanical properties in terms of tensile strength, ductility and fracture energy.
4 - The variations of ultimate tensile strength, ductility and fracture energy with volume fraction of martensite exhibit unusual behavior.
5 - Work hardening behavior was analyzed in terms of Holloman analysis. In this analysis, the work hardening behavior of the samples is single stage.

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